

## Intrinsic Mobility

$\mu$ : mobility

$$\mu = |v|/E > 0$$

$\mu$  is positive for both  $e^-$  and holes

$\mu_e, \mu_h$

The electric conductivity

$$\sigma = n e \mu_e + p e \mu_h$$

The drift velocity of a charge  $q$

$$v = q \frac{\tau E}{m}$$

$\tau$ : collision time

$m$ : effective mass

$$\mu_e = \frac{e \tau_e}{m_e}$$

$$\mu_h = \frac{e \tau_h}{m_h}$$

$\Rightarrow$  Mobility follows power law w.r.t. the  $T$  conductivity  $\rightarrow \propto \exp\left[-\frac{E_g}{2k_B T}\right]$

- mobility is limited by scattering with phonons
- Degeneracy of bands on the valence band limits the mobility  $\Rightarrow$  interband scattering

In ionic crystals holes are essentially immobile and can only hop from one ion to the next, provided they have enough thermal energy.

This self-trapping is due to the orbital degeneracy  $\Rightarrow$  most common in holes



## Impurity conductivity

Change the conductivity

In a Silicon (Si) lattice; if we replace some Si for Boron (B) atoms

1:  $10^5 \rightarrow$  cond. increases by  $10^3$

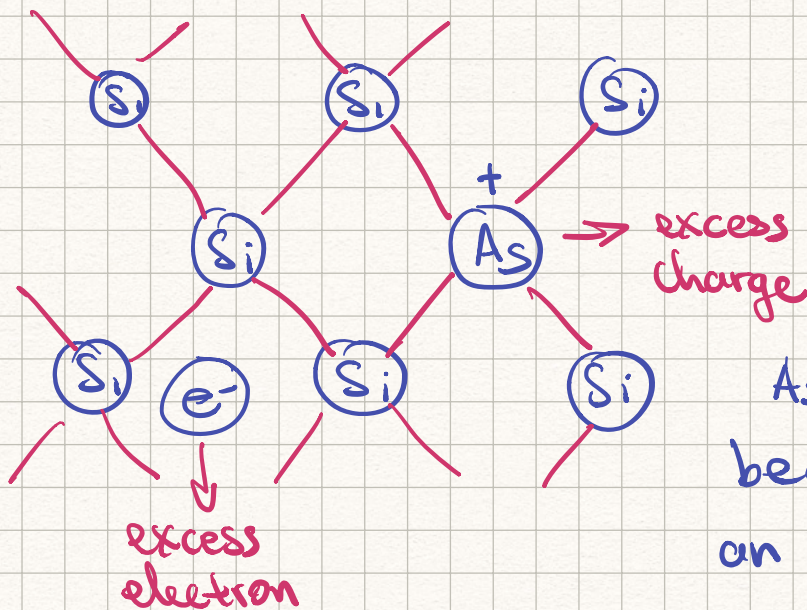
$\rightarrow$  Deliberately adding of impurities  $\rightarrow$  doping

## Donor states

$\rightarrow$  Si or Ge

- 4 valence electrons  $\rightarrow$  outer most shell
- Replace an atom Si or Ge with an atom of valence 5  
 $\rightarrow$  P, As, Sb

We will have a charge excess  $\rightarrow$  valence electron that is not paired



As gives  $1 e^-$  to the crystal  $\rightarrow$  DONOR

As is positively charged because it has lost an electron

The crystal remains neutral because the  $e^-$  remains in the crystal.



The extra  $e^-$  moves in a Coulomb potential

$$\frac{1}{4\pi\epsilon_0} \frac{e}{r} \frac{1}{\epsilon}$$

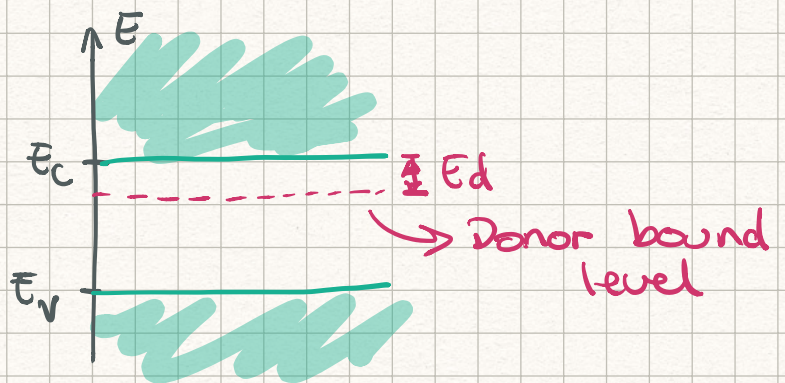
$\epsilon$ : dielectric constant of the crystal

$\frac{1}{\epsilon}$ : takes into account the reduction of the c. potential between charges.

Hydrogen atom:

• Ionization energy is  $-\frac{e^4 m}{2(4\pi\epsilon_0 \hbar)^2}$

•  $E_d = \frac{e^4 m_e}{2(4\pi\epsilon_0 \hbar)^2}$



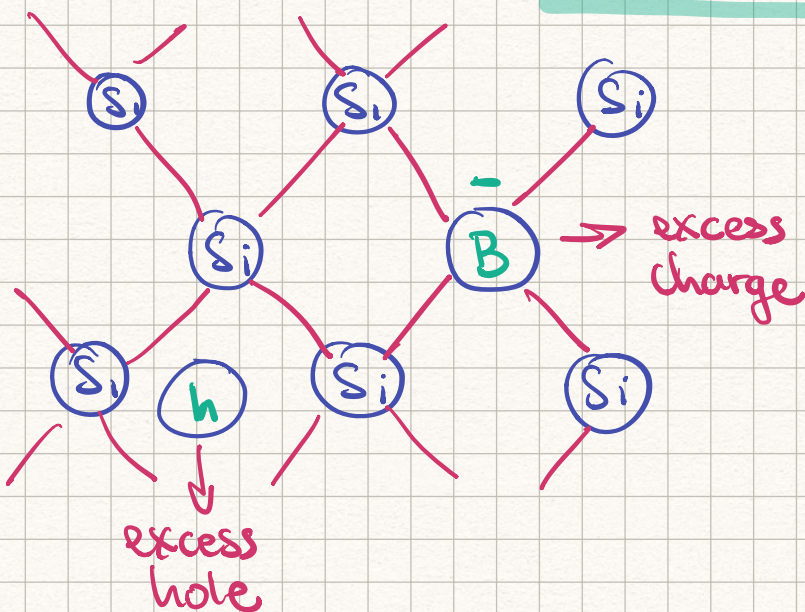
### Acceptor states

Replace an atom with another one with less valence electron

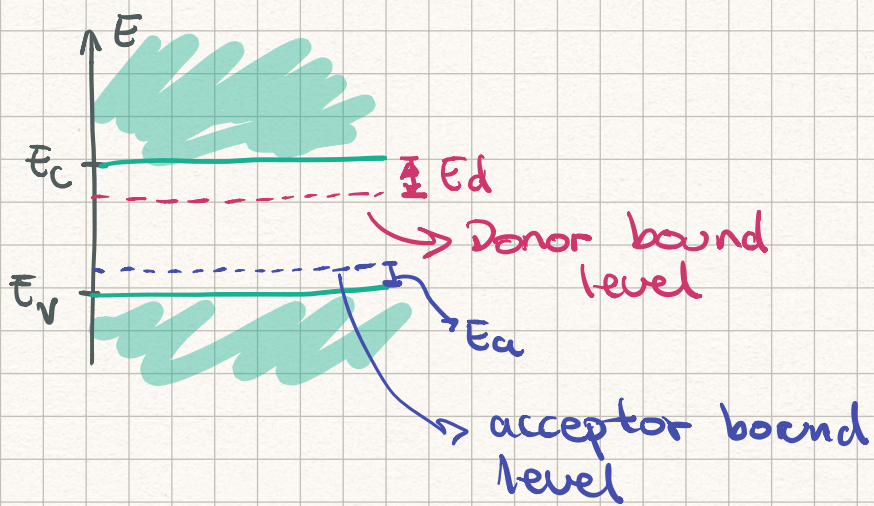
→ available to make bond

Si or Ge → B, Al, Ga, In

will need to take an  $e^-$  from the valence band to complete the bond







The Bohr model also applies to the holes, but the degeneracy complicates the problem

If

donor  $\gg$  acceptors : n-type doping semiconductor

donor  $\ll$  acceptors : p-type doping semiconductor

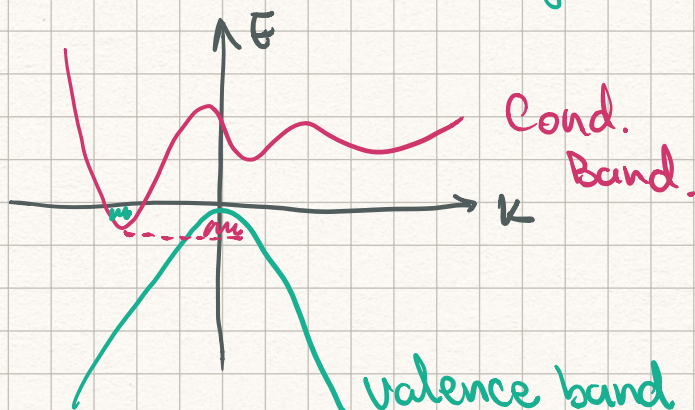
The Hall effect can be used to determine the type of semiconductor and the density of carriers

### Semimetal

$\rightarrow$  Arsenic (As); Antimony (Sb); Bismuth (Bi)  $\rightarrow$  group V

$\rightarrow$  Graphite

Conduction band edge is slightly below the valence band edge



The overlap in energy leads to a small amount of holes in the valence band and of  $e^-$  in the conduction band.

$\rightarrow$  They could be insulators

$\rightarrow$  pressure can change the overlap  $\rightarrow$  concentration of carriers